

MECHANOCHEMISTRY UNDER HYDROGEN GAS - SYNTHESIS OF HYDRIDES FOR ENERGY STORAGE

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Mechanochemistry under hydrogen gas is a method of choice for the synthesis of light-weight hydrides. In the last ten years, fast and efficient formation of key hydrides such as magnesium hydride (MgH_2)^[1,2], Mg-complex hydrides (Mg_2TMH_x , $\text{TM} = \text{Fe, Co, Ni}$)^[1], Mg-based nanocomposites ($\text{MgH}_2\text{-TiH}_2$, $\text{MgH}_2\text{-Si}$)^[3,4], alkali alanates (NaAlH_4 , KAlH_4)^[5,6], Li imide/amide ($\text{Li}_2\text{NH/LiNH}_2$)^[7] magnesium amide ($\text{Mg}(\text{NH}_2)_2$)^[8] and mixed-cation imide $\text{Li}_3\text{MgN}_2\text{H}$ ^[8] has been achieved in our laboratory. Besides being an elegant route for hydride formation, this method allows easy and controlled addition of dopants for fast screening studies^[9]. Moreover, reaction mechanisms during synthesis can be analyzed from the shape of in-situ hydrogenation curves. Last but not least, this method produces nanostructured materials with exceptional performances for hydrogen storage and electrochemical applications^[10-13]. This presentation honors the memory of our beloved colleague Michel Latroche who passed away recently and significantly contributed to this research topic.

References

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