

A NOVEL AMMINE LITHIUM ALUMINUM BOROHYDRIDE MATERIAL FOR HYDROGEN STORAGE

Huynh-Duc NGUYEN^a, Parviz HAJIYEV^a, Umit B. DEMIRCI^b, Gérard GEBEL^a

^a Univ Grenoble Alpes, CEA, Liten, DTNM, 38000 Grenoble, France

^b Institut Européen des Membranes, IEM – UMR 5635, ENSCM, CNRS, Univ Montpellier, Montpellier, France

E-mail: huynh-duc.nguyen@cea.fr

A novel solid ammine lithium aluminum borohydride $\text{LiAl}(\text{BH}_4)_4(\text{NH}_3)_6$ compound has the potential to be a good hydrogen storage material with 17.3 wt.% theoretical hydrogen capacity. It is synthesized at room temperature from LiAlH_4 and $(\text{CH}_3)_2\text{S-BH}_3$ precursors using the solvothermal method in liquid ammonia, which is a one-pot synthesis method that produces no solid byproducts or does not require the synthesis of pyrophoric compounds like $\text{Al}(\text{BH}_4)_3$.

X-ray powder diffraction analysis shows that the structure of $\text{LiAl}(\text{BH}_4)_4(\text{NH}_3)_6$, which has yet to be reported in the literature, does neither fit with known compounds such as $\text{Li}_2\text{Al}(\text{BH}_4)_5(\text{NH}_3)_6$ ¹ nor $\text{Al}(\text{BH}_4)_3(\text{NH}_3)_6$ ². Hence, the Rietveld refinement is performed to resolve the crystalline structure of this compound. Furthermore, FTIR, Raman, and MAS NMR spectroscopy are used to fully characterize $\text{LiAl}(\text{BH}_4)_4(\text{NH}_3)_6$. FTIR and Raman spectra reveal the presence of $[\text{BH}_4]^-$ and NH_3 groups, and the valence vibration of the Al-N bond in the coordination octahedron $[\text{Al}(\text{NH}_3)_6]^{3+}$. These findings are also supported by the ¹¹B and ²⁷Al MAS NMR spectra.

The dehydrogenation properties are investigated using TGA/DSC/MS/PCT characterization tools. The results show that this material decomposes exothermically at 154°C, releasing 12.5 wt.% H_2 with a purity of 96 mol% while the main contaminant being NH_3 . The hydrogen purity is improved to >99.0 mol% using the isothermal pretreatment method. In addition, using the Kissinger equation for kinetic analysis of dehydrogenation process, the activation energy is found to be 143.1 kJ/mol, which is comparable to similar compound such as ammonia borane³.

On another note, the amorphous Li-Al-B-N-H dehydrogenation product is unable to be rehydrogenated directly. This necessitates chemical regeneration. Hence, *in situ* XPS is being utilized to better understand the chemical composition and develop better chemical regeneration roots, much like its simpler synthesis process.

References

- (1) Guo, Y.; Wu, H.; Zhou, W.; Yu, X. Dehydrogenation Tuning of Ammine Borohydrides Using Double-Metal Cations. *J. Am. Chem. Soc.* **2011**, *133* (13), 4690–4693.
- (2) Guo, Y.; Yu, X.; Sun, W.; Sun, D.; Yang, W. The Hydrogen-Enriched Al–B–N System as an Advanced Solid Hydrogen-Storage Candidate. *Angew. Chem. Int. Ed.* **2011**, *50* (5), 1087–1091.
- (3) Gangal, A. C.; Sharma, P. Kinetic Analysis and Modeling of Thermal Decomposition of Ammonia Borane. *Int. J. Chem. Kinet.* **2013**, *45* (7), 452–461.

Picture of Author



Short Biography of Author

Huynh-Duc Nguyen is a PhD student at The French Alternative Energies and Atomic Energy Commission (CEA) since November 2020. He works on the hydrogen storage materials and their regeneration.

Huynh-Duc earned a degree in process engineering and a master's degree in chemistry, specialty of processes for valorization of renewable resources from the University of Technology of Compiègne (France), graduating in 2019. He currently lives in Grenoble.